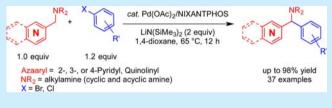


Palladium-Catalyzed Benzylic C—H Arylation of Azaarylmethylamines

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Supporting Information

ABSTRACT: A direct C-H functionalization approach to produce aryl(azaaryl)methylamines from azaarylmethylamines without directing groups is described. Under conditions where the azaarylmethylamines' C-H is reversibly deprotonated, a Pd(OAc)₂/NIXANTPHOS-based catalyst couples the resulting carbanions with various aryl halides to provide aryl-(azaaryl)methylamines. This umpolung strategy directly provides tertiary amines without protecting or activating groups.



yridine and quinoline derivatives are important heterocycles that are present in a broad range of biologically active natural products and pharmaceutically relevant compounds. Aryl(azaaryl)methylamines are prominent in biologically active small molecules, exhibiting antitumor,² antivirus,³ anti-HIV,⁴ and antihistamine activities⁵ (Figure 1). Furthermore, recent statistical studies indicate that the prominence of heteroaromatic rings in marketed oral drugs is increasing.6

Farnesyl Protein Transferase Inhibitor Enterovirus Inhibitor (X = N or CH)HIV Integrase Inhibitor Hypolipidemic Activity

Figure 1. Selected pharmacologically active compounds containing aryl(azaaryl)methylamines.

Conventional approaches to the synthesis of 1,1-diarylmethylamine derivatives⁷ include substitution reactions with amine nucleophiles, addition of aryl or azaaryl organometallic reagents to imines, and reduction of imines.8 Despite the popularity of these methods, they involve prefunctionalized electrophiles or nucleophiles. Transition-metal-catalyzed crosscoupling reactions also represent an attractive approach.^{7d} The

direct C-H functionalization of amino alkyl moieties (R'CH₂NR₂), however, is challenging. Therefore, activated substrates, such as Boc-protected amines, ketimines, and (η^6 benzylamine)Cr(CO)₃ complexes, have been employed to facilitate deprotonation of C-H bonds adjacent to nitrogen (Figure 2). The activating groups necessary in Figure 2 to

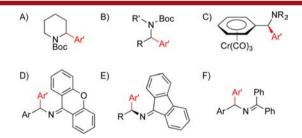


Figure 2. Representative synthetic intermediates with directing groups (A, B) and activating groups (C-F) employed in metal-catalyzed C-H functionalization adjacent to nitrogen.

increase reactivity are often subsequently transformed into the desired functional groups or excised from the product entirely, decreasing the synthetic efficiency.

Considering the importance of aryl(azaaryl)methylamines, the development of in situ deprotonation/arylation strategies for their synthesis that do not employ activating groups would be valuable. Based on our experience in deprotonative cross-coupling reactions $^{10c-f,11}$ with weakly acidic substrates, we hypothesized that the reversible deprotonation of (aminomethyl)pyridines in the presence of a palladium catalyst should be possible.

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Herein, we report a practical method to prepare aryl-(azaaryl)methylamines from (aminomethyl)pyridines and aryl halides using a Pd—NIXANTPHOS-based catalyst (Scheme 1).

Scheme 1. Palladium-Catalyzed Benzylic C-H Arylation with Azaarylmethylamines

This approach affords single-step access to aryl(azaaryl)-methylamines from simple starting materials. Notably, to the best of our knowledge, the direct functionalization of aminomethyl azaarenes has not been successfully achieved prior to this work.

We anticipated two challenges to advancing a successful deprotonative cross-coupling process (DCCP) with (aminomethyl)pyridine derivatives: (1) introduction of conditions for the in situ deprotonation of the weakly acidic C-H's of the substrate and (2) identification of a catalyst capable of promoting coupling of the resulting organolithium, -sodium, or -potassium derivatives with aryl halides. In the latter case, we have found that palladium complexes of van Leeuwen's NIXANTPHOS ligand¹² outperform other Pd-phosphine complexes in this class of reactions. As such, it has become our "go to" ligand for new substrate classes. Concerning the deprotonation step, the p K_a of the sp^3 -hybridized C-H's of 2-(aminomethyl)pyridines are unknown, although that of 2methylpyridine has been determined to be 34 in THF. 13 The high pK, value of these substrates suggests that alkoxide and silylamide bases are good starting points. Additionally, palladium-catalyzed benzylic arylations of pyridine derivatives have become a powerful method to construct aryl(azaaryl)methane derivatives. 14

We initially tested the reactions of 2-(morphorinomethyl)-pyridine (1a) with 1-bromo-4-tert-butylbenzene using Pd-(OAc)₂ (5 mol %) and NIXANTPHOS (7.5 mol %) in THF at 65 °C. Six bases [LiO-t-Bu, NaO-t-Bu, KO-t-Bu, LiN-(SiMe₃)₂, NaN(SiMe₃)₂, and KN(SiMe₃)₂] were employed, and the results are displayed in entries 1–6 of Table 1.

Although MO-t-Bu bases [M = Li, Na, K] did not generate the desired product 3a (entries 1–3), $MN(SiMe_3)_2$ bases (M = Li, Na, K) were all very promising, affording 80-87% assay yield (AY) of the cross-coupling product 3a (entries 4-6). The most promising result was obtained with LiN(SiMe₃)₂ (87% AY as determined by ¹H NMR, entry 4). We next screened other ethereal solvents (DME, CPME, 1,4-dioxane) (entries 7-9). Of these, 1,4-dioxane led to 90% AY of the desired product (entry 9). Examination of the stoichiometry indicated use of a 1:1.2:2 ratio of pyridylmethylamine (1a)/1-bromo-4-tertbutylbenzene (2a)/LiN(SiMe₃)₂ rendered 98% AY of 3a, leading to 92% isolated yield (entry 11). Use of a stoichiometric amount of 1-bromo-4-tert-butylbenzene (2a) or lower catalyst loading [2.5 mol % Pd(OAc)₂ and 3.75 mol % of NIXANTPHOS] resulted in incomplete conversion at 65 °C after 12 h (entries 12-13). The optimized conditions (entry 11, Table 1) were then used to define the substrate scope.

The scope of the arylation was initially explored with 2-(cyclic amino)methylpyridines (1) using aryl bromides

Table 1. Selected Optimization of Pd-Catalyzed $C(sp^3)$ -H Arylation of 1a with 2a

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(N)	Br +	Pd(OAc) ₂ (5 mo NIXANTPHOS (7.5		N
CN	tBu	base solvent 65 °C, 12 h		/Bu
1a	2a			3a
entry	base	1a/2a/base	solvent ^a	yield b (%)
1	LiO-t-Bu	1:1.5:3	THF	0
2	NaO-t-Bu	1:1.5:3	THF	0
3	KO-t-Bu	1:1.5:3	THF	0
4	$LiN(SiMe_3)_2$	1:1.5:3	THF	87
5	$NaN(SiMe_3)_2$	1:1.5:3	THF	80
6	$KN(SiMe_3)_2$	1:1.5:3	THF	83
7	$LiN(SiMe_3)_2$	1:1.5:3	DME	82
8	$LiN(SiMe_3)_2$	1:1.5:3	CPME	85
9	$LiN(SiMe_3)_2$	1:1.5:3	1,4-dioxane	90
10	$LiN(SiMe_3)_2$	1:1.5:2	1,4-dioxane	92
11	$LiN(SiMe_3)_2$	1:1.2:2	1,4-dioxane	98 (92) ^c
12	$LiN(SiMe_3)_2$	1:1:1.5	1,4-dioxane	73 ^d
13	$LiN(SiMe_3)_2$	1:1.2:2	1,4-dioxane	64 ^e

^aTetrahydrofuran (THF), 1,2-dimethoxyethane (DME), cyclopentyl methyl ether (CPME), and 1,4-dioxane (dioxane). ^bYields were determined by ¹H NMR analysis of unpurified reaction mixtures with internal standard CH₂Br₂. ^cIsolated yield. ^d14% of 1a remained. ^e2.5 mol % of Pd(OAc)₂ and 3.75 mol % of NIXANTPHOS.

(Scheme 2). Morpholine, pyrrolidine, and N-methylpiperazine were coupled with aryl bromides bearing various substituents, generating aryl(2-pyridyl)methylamines in good to excellent isolated yields (65-98%). The arylated products 3b-d were obtained in 78-83% yield with bromobenzene and 1bromonaphthalene. Under the same reaction conditions, chlorobenzene was also a suitable coupling partner, providing 3b in 78% yield. Aryl bromides bearing electron-donating 4-NMe₂ and 4-OMe resulted in coupling products 3e-h in 71-94% yield. Aryl bromides with electron-withdrawing 4-F and 3-CF₃ groups were well tolerated and provided aryl(2-pyridyl)methylamines 3i-k in 71-94% yield. Heterocyclic 3bromopyridine was also a suitable coupling partner to generate dipyridylmethyl morpholine 31 in 65% yield. The reactions with 1 mol % of catalyst generally occurred in good yields (71–75% yield, 3d, 3g, and 3i). To illustrate the scalability of our method, we conducted coupling of 2-(morphorinomethyl)pyridine (1a, 10 mmol) with 1-bromo-4-fluorobenzene using 4 mol % of catalyst. The coupling product 3m was isolated in 96% yield (2.61 g).

Next, we turned our attention to acyclic amines. The optimized arylation conditions employed with cyclic amines were easily transferable to *N,N*-dimethyl- or -diethylamine derivatives with a variety of aryl bromides. Thus, aryl bromides with neutral (4-*t*-Bu, 3-Me), electron-donating (4-OMe, 4-NMe₂), or electron-withdrawing (4-F, 4-Cl, 3-OMe, 3-CF₃) substituents coupled with 2-(*N,N*-dialkylaminomethyl)pyridine, providing the cross-coupling products 4a—o in 60—98% yield (Scheme 3). The yields with *N,N*-dimethylamine derivatives are slightly higher than those with cyclic amines. Several relatively challenging substrates (1-bromo-4-trifluoromethylbenzene and 5-bromobenzofuran) were also competent coupling partners, albeit in reduced yields (4i,j, 60—66% yield). While 1-bromonaphthylene furnished coupling product 4k in 69%

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Scheme 2. Pd-Catalyzed $C(sp^3)$ -H Arylation of 2-Pyridylmethylamines 1 with Aryl Bromides 2^a

"Isolated yields. "X = Br was used unless noted. "5 mol % of Pd(OAc)₂ and 7.5 mol % of NIXANTPHOS and the yields in parentheses with 1 mol % of Pd(OAc)₂ and 1.5 mol % of NIXANTPHOS. "4 mol % of Pd(OAc)₂ and 6 mol % of NIXANTPHOS.

Scheme 3. Benzylic Cross-Coupling of 2-(Dialkylaminomethyl)pyridine 1 with Aryl Bromides 2^a

yield, the sterically more hindered 2-bromotoluene failed to provide the desired product (<2% yield). 2-(3-Bromophenyl)-1,3-dioxolane generated acetal-protected aldehyde product 4l in 60% yield. When the same conditions were applied to 2-(N,N-diethylaminomethyl)pyridine, the desired products $4\mathbf{m} - \mathbf{o}$ were

isolated in 66-69% yield. Sterically more hindered 2-(N-benzyl-N-methylaminomethyl)pyridine did not react (<2% yield). Similar results were observed in related reactions by Baudoin and our groups with bulkier substituents on nitrogen. 110,15

The above results led us to wonder if this method would be limited to 2-pyridyl derivatives because it requires a directed metalation for the deprotonation. To answer this question, other heterocyclic substrates were tested. Although the pK_a 's of 4-pyridylmethylamines are not reported, 4-methylpyridine has a pK_a of 32.2 in THF.¹³ As shown in Scheme 4, 4-pyridylmethyl-

Scheme 4. Benzylic Cross-Coupling of Azaarene Derivatives

"Isolated yields. bX = Br unless noted. c1 mol % of Pd(OAc)₂ and 1.5 mol % of NIXANTPHOS and the yields in parentheses with 5 mol % of Pd(OAc)₂ and 7.5 mol % of NIXANTPHOS. d5 mol % of Pd(OAc)₂ and 10 mol % of NIXANTPHOS. e The reaction was conducted in THF at 110 °C with KN(SiMe₃)₂ instead of LiN(SiMe₃)₂.

amine derivatives were excellent substrates, providing coupling products in 55-92% yield (5a-h) with 1 mol % catalyst loading. These results indicate that chelation of the base to the 2-pyridyl group is not necessary for the reversible deprotonation step of this DCCP.

Additional heterocycles were also amenable to the DCCP. Under the conditions used for 2-pyridylamines, a 2-(aminomethyl)quinoline derivative underwent a direct arylation reaction in 52% yield (5i). The least acidic substrate tested in this study was the 3-pyridylmethylamine derivative. The p K_a of 3-methylpyridine is 37.7 in THF. ¹³ After extensive optimization of bases, solvents, and temperature, we were able to achieve coupling in the presence of 2 equiv of KN(SiMe₃)₂ in THF at 110 °C, resulting in arylation product 5j in 65% yield.

In summary, we report the first direct palladium-catalyzed benzylic $C(sp^3)$ —H arylation of cyclic and acyclic aminomethyl-substituted azaarenes with aryl bromides. A variety of aryl(azaaryl)methylamines were prepared in good to excellent yields from readily accessible tertiary aminomethyl azaarenes. It is noteworthy that addition and removal of directing groups was not necessary with these substrates, facilitating their streamlined synthesis. This new method enables the rapid preparation of druglike molecules using straightforward

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techniques that are ideal for applications in medicinal chemistry. 16

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02898.

Spectroscopic characterization data and procedures for preparation of all new compounds (PDF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) (a) Henry, G. D. Tetrahedron 2004, 60, 6043. (b) Michael, J. P. Nat. Prod. Rep. 2005, 22, 627. (c) Laird, T. Org. Process Res. Dev. 2006, 10, 851. (d) Joule, J. A.; Mills, K. Heterocyclic Chemistry, 5th ed.; John Wiley & Sons: Chichester, U.K., 2010. (e) Alexander, F. Pozharskii, A. S., Katritzky, A. R. Heterocycles in Life and Society: An Introduction to Heterocyclic Chemistry, Biochemistry and Applications, 2nd ed.; John Wiley & Sons: Chichester, U.K., 2011. (f) Baumann, M.; Baxendale, I. R. Beilstein J. Org. Chem. 2013, 9, 2265.
- (2) Liu, M.; Bryant, M. S.; Chen, J.; Lee, S.; Yaremko, B.; Li, Z.; Dell, J.; Lipari, P.; Malkowski, M.; Prioli, N.; Rossman, R. R.; Korfmacher, W. A.; Nomeir, A. A.; Lin, C. C.; Mallams, A. K.; Doll, R. J.; Catino, J. J.; Girijavallabhan, V. M.; Kirschmeier, P.; Bishop, W. R. Cancer Chemother. Pharmacol. 1998, 43, 50.
- (3) Chern, J.-H.; Shia, K.-S.; Hsu, T.-A.; Tai, C.-L.; Lee, C.-C.; Lee, Y.-C.; Chang, C.-S.; Tseng, S.-N.; Shih, S.-R. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 2519.
- (4) Summa, V.; Petrocchi, A.; Matassa, V. G.; Gardelli, C.; Muraglia, E.; Rowley, M.; Paz, O. G.; Laufer, R.; Monteagudo, E.; Pace, P. *J. Med. Chem.* **2006**, *49*, 6646.
- (5) Anagnostopulos, H.; Bartlett, R. R.; Elben, U.; Stoll, P. Eur. J. Med. Chem. 1989, 24, 227.
- (6) Ritchie, T. J.; Macdonald, S. J. F.; Young, R. J.; Pickett, S. D. Drug Discovery Today 2011, 16, 164.
- (7) (a) Doggrell, S. A.; Liang, L. C. Naunyn-Schmiedeberg's Arch. Pharmacol. 1998, 357, 126. (b) Plobeck, N.; Delorme, D.; Wei, Z.-Y.; Yang, H.; Zhou, F.; Schwarz, P.; Gawell, L.; Gagnon, H.; Pelcman, B.; Schmidt, R.; Yue, S. Y.; Walpole, C.; Brown, W.; Zhou, E.; Labarre, M.; Payza, K.; St-Onge, S.; Kamassah, A.; Morin, P.-E.; Projean, D.; Ducharme, J.; Roberts, E. J. Med. Chem. 2000, 43, 3878. (c) Ko, Y.; Malone, D. C.; Armstrong, E. P. Pharmacotherapy 2006, 26, 1694. (d) Ameen, D.; Snape, T. J. MedChemComm 2013, 4, 893.
- (8) (a) Kobayashi, S.; Ishitani, H. Chem. Rev. 1999, 99, 1069. (b) Carey, F. A.; Sundberg, R. J. Advanced Organic Chemistry: Part B: Reaction and Synthesis, 5th ed.; Springer: New York, 2007. (c) Kobayashi, S.; Mori, Y.; Fossey, J. S.; Salter, M. M. Chem. Rev. 2011, 111, 2626. (d) Smith, M. B.; March, J. March's Advanced Organic Chemistry: Reactions, Mechanisms, and Structure, 7th ed.; John Wiley & Sons, Inc.: Hoboken, 2013.

- (9) Dastbaravardeh, N.; Schnuerch, M.; Mihovilovic, M. D. Org. Lett. 2012, 14, 1930.
- (10) (a) Niwa, T.; Yorimitsu, H.; Oshima, K. Org. Lett. 2008, 10, 4689. (b) Niwa, T.; Suehiro, T.; Yorimitsu, H.; Oshima, K. Tetrahedron 2009, 65, 5125. (c) McGrew, G. I.; Temaismithi, J.; Carroll, P. J.; Walsh, P. J. Angew. Chem., Int. Ed. 2010, 49, 5541. (d) McGrew, G. I.; Stanciu, C.; Zhang, J.; Carroll, P. J.; Dreher, S. D.; Walsh, P. J. Angew. Chem., Int. Ed. 2012, 51, 11510. (e) Li, M.; Yücel, B.; Adrio, J.; Bellomo, A.; Walsh, P. J. Chem. Sci. 2014, 5, 2383. (f) Li, M.; Berritt, S.; Walsh, P. J. Org. Lett. 2014, 16, 4312. (g) Fernandez-Salas, J. A.; Marelli, E.; Nolan, S. P. Chem. Sci. 2015, 6, 4973.
- (11) (a) Zhang, J.; Bellomo, A.; Creamer, A. D.; Dreher, S. D.; Walsh, P. J. J. Am. Chem. Soc. 2012, 134, 13765. (b) Jia, T.; Bellomo, A.; El Baina, K.; Dreher, S. D.; Walsh, P. J. J. Am. Chem. Soc. 2013, 135, 3740. (c) Montel, S.; Jia, T.; Walsh, P. J. Org. Lett. 2014, 16, 130. (d) Zheng, B.; Jia, T.; Walsh, P. J. Org. Lett. 2013, 15, 1690. (e) Zheng, B.; Jia, T. Z.; Walsh, P. J. Org. Lett. 2013, 15, 4190. (f) Frensch, G.; Hussain, N.; Marques, F. A.; Walsh, P. J. Adv. Synth. Catal. 2014, 356, 2517. (g) Gao, F.; Kim, B.-S.; Walsh, P. J. Chem. Commun. 2014, 50, 10661. (h) Hussain, N.; Frensch, G.; Zhang, J.; Walsh, P. J. Angew. Chem., Int. Ed. 2014, 53, 3693. (i) Montel, S.; Raffier, L.; He, Y.; Walsh, P. J. Org. Lett. 2014, 16, 1446. (j) Yücel, B.; Walsh, P. J. Adv. Synth. Catal. 2014, 356, 3659. (k) Zheng, B.; Jia, T.; Walsh, P. J. Adv. Synth. Catal. 2014, 356, 165. (1) Zhang, J.; Bellomo, A.; Trongsiriwat, N.; Jia, T.; Carroll, P. J.; Dreher, S. D.; Tudge, M. T.; Yin, H.; Robinson, J. R.; Schelter, E. J.; Walsh, P. J. J. Am. Chem. Soc. 2014, 136, 6276. (m) Mao, J.; Eberle, K.; Zhang, J.; Rodríguez-Escrich, C.; Xi, Z.; Pericàs, M. A.; Walsh, P. J. Tetrahedron Lett. 2015, 56, 3604. (n) Sha, S.-C.; Zhang, J.; Walsh, P. J. Org. Lett. 2015, 17, 410. (o) Hussain, N.; Kim, B.-S.; Walsh, P. J. Chem. - Eur. J. 2015, 21, 11010.
- (12) (a) van der Veen, L. A.; Keeven, P. H.; Schoemaker, G. C.; Reek, J. N. H.; Kamer, P. C. J.; van Leeuwen, P. W. N. M.; Lutz, M.; Spek, A. L. *Organometallics* **2000**, *19*, 872. (b) Kamer, P. C. J.; van Leeuwen, P. W. N. M.; Reek, J. N. H. *Acc. Chem. Res.* **2001**, *34*, 895. (13) Fraser, R. R.; Mansour, T. S.; Savard, S. *J. Org. Chem.* **1985**, *50*, 3232.
- (14) (a) Niwa, T.; Yorimitsu, H.; Oshima, K. Angew. Chem., Int. Ed. 2007, 46, 2643. (b) Niwa, T.; Yorimitsu, H.; Oshima, K. Org. Lett. 2007, 9, 2373. (c) Hlavinka, M. L.; Hagadorn, J. R. Organometallics 2007, 26, 4105. (d) Campeau, L.-C.; Schipper, D. J.; Fagnou, K. J. Am. Chem. Soc. 2008, 130, 3266. (e) Mousseau, J. J.; Larivée, A.; Charette, A. B. Org. Lett. 2008, 10, 1641. (f) Schipper, D. J.; Campeau, L.-C.; Fagnou, K. Tetrahedron 2009, 65, 3155. (g) Burton, P. M.; Morris, J. A. Org. Lett. 2010, 12, 5359. (h) Shang, R.; Yang, Z.-W.; Wang, Y.; Zhang, S.-L.; Liu, L. J. Am. Chem. Soc. 2010, 132, 14391. (i) Duez, S.; Steib, A. K.; Manolikakes, S. M.; Knochel, P. Angew. Chem., Int. Ed. 2011, 50, 7686. (j) Song, G.; Su, Y.; Gong, X.; Han, K.; Li, X. Org. Lett. 2011, 13, 1968. (k) Zhao, D.; Zhu, M.-X.; Wang, Y.; Shen, Q.; Li, J.-X. Org. Biomol. Chem. 2013, 11, 6246.
- (15) Millet, A.; Dailler, D.; Larini, P.; Baudoin, O. Angew. Chem., Int. Ed. 2014, 53, 2678.
- (16) From preliminary experiments screening 37 different Ni(phosphine)-based catalysts in the coupling of **1a** with bromobenzene, it was found that the Ni(NIXANTPHOS)-based catalyst gave the highest product/internal standard ratio; see: Cao, X.; Sha, S.-C.; Li, M.; Kim, B.-S.; Morgan, C.; Huang, R.; Yang, X.; Walsh, P. J. Chem. Sci. **2016**, DOI: 10.1039/C5SC03704B.